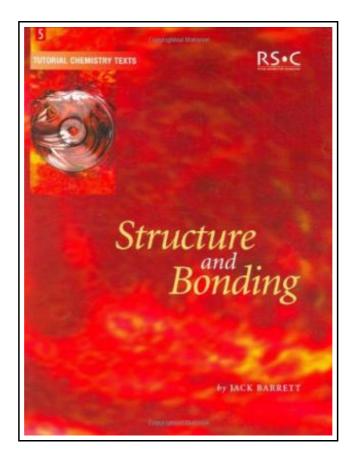
# Structure and Bonding (Paperback)



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## Reviews

Completely essential read pdf. It is definitely simplistic but shocks within the 50 % of your book. Its been designed in an exceptionally straightforward way which is simply following i finished reading through this publication in which actually changed me, change the way i believe. (Damon Friesen)

# STRUCTURE AND BONDING (PAPERBACK)



Royal Society Of Chemistry, United Kingdom, 2001. Paperback. Book Condition: New. 244 x 184 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*. Structure and Bonding covers introductory atomic and molecular theory as given in first and second year undergraduate courses at university level. This book explains in non-mathematical terms where possible, the factors that govern covalent bond formation, the lengths and strengths of bonds and molecular shapes. Throughout the book, theoretical concepts and experimental evidence are integrated. An introductory chapter summarizes the principles on which the Periodic Table is established, and describes the periodicity of various atomic properties which are relevant to chemical bonding. Symmetry and group theory are introduced to serve as the basis of all molecular orbital treatments of molecules. This basis is then applied to a variety of covalent molecules with discussions of bond lengths and angles and hence molecular shapes. Extensive comparisons of valence bond theory and VSEPR theory with molecular orbital theory are included. Metallic bonding is related to electrical conduction and semi-conduction. The energetics of ionic bond formation and the transition from ionic to covalent bonding is also covered. Ideal for the needs of undergraduate chemistry students, Tutorial Chemistry Texts is a major series consisting of short, single topic or modular texts concentrating on the fundamental areas of chemistry taught in undergraduate science courses. Each book provides a concise account of the basic principles underlying a given subject, embodying an independent-learning philosophy and including worked examples.



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